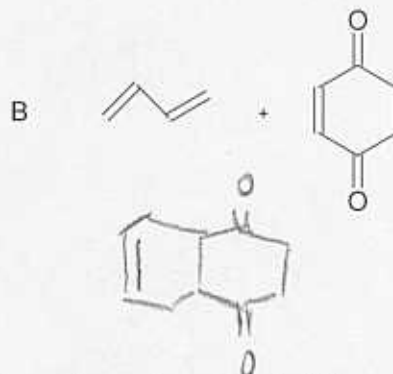
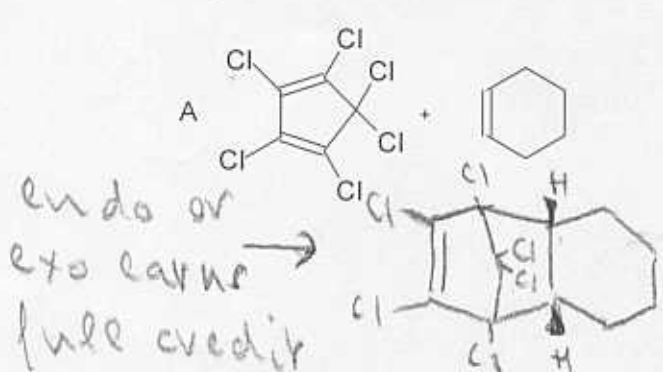


1. Show valid structural formulas for the products of the following Diels-Alder reactions (4 pts)



2. Is reaction A above a normal or reverse electron demand reaction? What about reaction B? (2 pts)

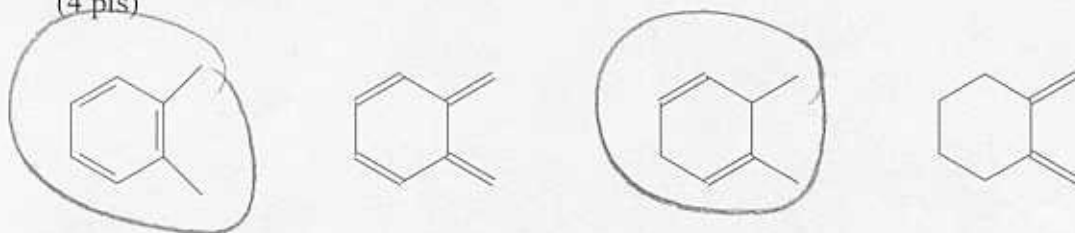
A - reverse

B - normal

3. Circle any CORRECT statements (4 pts)

- a. Diels-Alder reactions work best if both the diene and the dienophile are electron-rich.
- b. All Diels-Alder adducts must contain a six-membered ring.
- c. Diels-Alder reactions work best if both the diene and the dienophile are electron-depleted.
- d. The xylene used in last week's experiment acts as a catalyst.

4. Circle all that are UNABLE to serve as the diene component in Diels-Alder reactions. (4 pts)



5. Show a valid structural formula for *ortho*-xylene! (1 pts)

